

Microwave-Assisted Synthesis of Substituted Pyrrolo[2,3-d]pyrimidines

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37 A new synthetic route to triaryl pyrrolo[2,3-d]pyrimidines from common 4,6-dichloropyrimidine has
38 been developed. The triarylated compounds are synthesized by three crosscoupling reactions using three
39 different catalysts. The introduction of a C-6 aryl group was achieved in a two-step process under
40 Sonogashira conditions [Pd(dba)₂/CuI] followed by intramolecular cyclization, and application of
41 Suzuki–Miyaura conditions [Pd(PPh₃)₄; PdCl₂(PPh₃)₂] led to C-4 and C-5 diarylation. This sequence
42 allows a flexible synthetic approach to highly arylated pyrrolopyrimidines containing different aryl
43 groups.

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INTRODUCTION

Pyrrolo[2,3-d]pyrimidines (IUPAC numbering is used throughout the manuscript), also known as 7-deazapurines, are an important class of heterocyclic compound containing a pyrrole ring with pyrimidine fused at the α,β -position. The pyrrolo[2,3-d]pyrimidine core is found in a wide range of natural compounds, including nucleoside antibiotics such as toyocamycin, sangivamycin and tubercidin.[1] Its wide and versatile spectrum of biological activities has led to its incorporation in synthetic biologically active compounds such as neurogenesis inducers by GSK-3 β inhibition (TWS119, Figure 1),[2] antitumor (ACK1 inhibitors with IC₅₀ = 0.62 μ m, Figure 1),[3] anti-inflammatory agents,[4] and analgesic derivatives.[5] In addition to these bioapplications, other structurally diverse substituted purines[6] and 7-deazapurines[7,8] have been studied for their photophysical properties as, for example, electronic material and blue-light emitters. In recent years, great efforts have been dedicated to the synthesis of this heterocyclic system and to the development and functionalization of biologically active compounds containing the pyrrolo[2,3-d]pyrimidine substructure.[9]

General methods for the preparation of pyrrolopyrimidines remain limited.[10] The majority of procedures provide C-4 monosubstituted or C-2, N-7 disubstituted compounds and there have been few general approaches to the functionalization of this heterocycle system.[11] In general, compounds possessing this nitrogenated scaffold can be prepared from alicyclic reagents in basic media.[12] A synthesis of pyrrolo[2,3-d]pyrimidines by the cyclization of appropriately substituted pyrrole was reported by Rashad et al.[13] and Hilmy et al.[14] The standard method entails the introduction of the pyrrole ring to bridge the pyrimidine at the C-4, C-5 positions. Other previously published synthetic approaches to pyrrolopyrimidines have used halogenated pyrimidines as the common intermediate, which are converted into the corresponding aminopyrimidines.[15] The crosscoupling reaction of aminopyrimidines with vinyl stannanes followed by cyclization furnishes the pyrrolopyrimidines without functionalization of the five-membered ring.[15] The reaction of 4-amino-5-bromopyrimidine through ketone α -arylation with acetophenone provides the 6-phenylpyrrolopyrimidine in 62% yield.[16] Recently, Kopecky and coworkers[3] reported a general route with which to access the bicyclic structure from 4,6-diamino-5-iodopyrimidine involving an intramolecular palladium-catalyzed Heck reaction. The cyclization reaction of 2-alkynylanilines constitutes an interesting procedure with which to obtain 2-substituted indoles, azaindoles or other heterocyclic cores including pyrrolopyrimidines.[17] However, this intramolecular cyclization requires a base and high temperatures (180–200 °C).[17,18]

Despite the wide range of methods available for the preparation of pyrrolopyrimidines, very few of them provide an efficient procedure with which to obtain the heterocyclic system with a range of substituents. To the best of our knowledge, the limited number of approaches available for the synthesis and functionalization of pyrrolo[2,3-d]pyrimidines do not include the preparation of 4,5,6-triarylpyrrolopyrimidines. Only 2,4,7-triarylpyrrolo[2,3-d]pyrimidines[19] and 2,6,8-triarylpurines,[20] possessing optical or adenosine antagonist properties, respectively, have been reported.

RESULTS AND DISCUSSION

With the aim of developing an efficient route that is suitable for the triarylation of pyrrolopyrimidines, we designed the following general synthesis from the commercially available precursor 4,6-dichloropyrimidine (1; Scheme 1). This synthesis was performed in seven steps by using microwave irradiation under mild to moderate conditions from stable starting material.

The starting 5-alkynylpyrimidines 4 can be prepared easily by a Sonogashira type alkynylation[21] of 6-chloro-5-iodo-4-(methylamino)pyrimidine by using commercial alkynylbenzenes under microwave assistance. The use of microwave irradiation under palladium catalysis has been investigated and applied to organic synthesis in both academia and industry.[22] The Sonogashira reaction is chemoselective and the yield is not affected by the electronic properties of the 4-substituted aryl alkyne (Scheme 2). The 6-chloro-5-iodo-4-(methylamino)pyrimidine (3) used in this reaction was prepared by nucleophilic substitution of the 4,6-dichloropyrimidine with methylamine at C-4 followed by iodination under classical conditions.[17]

The intramolecular cyclization in the presence of base (Cs_2CO_3) and a catalytic amount of CuI (1 mol-%) under microwave irradiation afforded bicyclic heterocycle 5 in excellent yield with good tolerance of different substituent groups. These conditions allowed us to work at lower temperatures (100 °C) than those published by other authors (180–200 °C).[17,18] Applying the same conditions but performing the cyclization without CuI provided higher yields, showing that the reaction works even better without the copper catalyst (Scheme 3). The optimized protocol involved treating diarylalkyne with cesium carbonate in acetonitrile and heating at 100 °C for 30–90 min in a microwave system. Ring closure was achieved to give the desired product 5 in only one step from the common intermediate 4; the product was used in the next step without further purification.

In a preliminary study, the direct halogenation of 4-chloropyrrolopyrimidines 5 by using N-bromosuccinimide (NBS) or N-iodosuccinimide (NIS), under microwave assistance, provided the corresponding 5-bromopyrrolopyrimidines (87%) or 5-iodopyrrolopyrimidines (90%) (Scheme 4). Replacement of the bromo or iodo group attached at the C-5 position of the pyrrolopyrimidines by using the corresponding arylboronic acid provided the arylated compounds 9 in yields of 33 or 70%, from the bromo or iodo derivative, respectively.

In the following step, the isolated 4-chlorodiarylated pyrrolopyrimidines 9 were arylated by using aryl boronic acids and Suzuki–Miyaura catalysts in 83% yield. This synthetic procedure furnished triarylpyrrolopyrimidines 8 in seven steps with an overall yield of 12–32%.

With these results in hand, we attempted to generate triarylpyrrolopyrimidines from intermediate 5 by using an alternative route, reversing the order of arylation of the pyrrolopyrimidines so that arylation would first take place at C-4 and then at C-5. Thus, 4-aryl derivatives 6 were obtained by treatment of 4-chloropyrrolopyrimidines 5 with arylboronic acids under Suzuki–Miyaura cross-coupling conditions[23] by using a microwave heat source.[24] The reaction yields were good to excellent (91–100%; Table 1), and the reaction incorporating the phenyl derivatives was found to be chemoselective. This method thus provides efficient access to diarylated compounds.

Direct arylation of C-5 was not achieved under a wide range of tested conditions. As a result, the diarylpyrrolopyrimidines were iodinated at the C-5 position by electrophilic substitution with N-iodosuccinimide[25] under microwave radiation at 100 °C with nearly quantitative conversion, very high yields, and sufficient purity as to obviate the need for a purification step (Table 2). Halogenation at C-5 was performed regioselectively. Interestingly, analysis of the iodo compounds by ^1H NMR spectroscopy carried out at 30, 100, and 180 days after their preparation showed higher stability at room temperature when the compounds were stored without a solvent. Following this study, the iodo

compounds in chloroform solution showed complete stability after 30 days at room temperature. It should be emphasized that carrying out the C-4 arylation of the pyrrolopyrimidines before C-5 iodination avoided the Cl/I exchange that impedes the chemoselective arylation.

The third aryl group at the C-5 position of 4,6-diaryl-5-iodopyrrolo[2,3-d]pyrimidines **7** was introduced by treatment with arylboronic acids under Suzuki–Miyaura cross-coupling conditions. After workup, a new iodination process with NIS was carried out, which allowed some additional starting material to be recovered, separated, and reused. Microwave assistance enabled a rapid synthesis of a variety of triaryl pyrrolopyrimidines with different substituents. It was clear that meta and para substituents on the arylboronic acid (Ar²) were well-tolerated, whereas ortho substituents hampered the reaction as a result of steric hindrance (Table 3, entry 5). The ortho, meta or para substituents of the aryl groups attached at C-4 of the pyrrolopyrimidine core did not modify the arylation process (Table 3, entries 1, 6 and 9). Whereas the presence of a substituent of the aryl group linked to C-6 (Table 3, cf. entry 1 with entries 10 and 11) had a significant impact on the rate of the Suzuki–Miyaura reaction, the electron-donating or withdrawing nature of the substituents was not influential. Increasing the number of equivalents of the reactant, as well as the reaction time and temperature did not favor the reaction. The optimized procedure afforded the desired triaryl-substituted pyrrolopyrimidines in 35–50% overall yield in seven steps, which constitutes a significant improvement on the previously described route.

The molecular structure of the triaryl-substituted pyrrolo[2,3-d]pyrimidine **8k** (Table 3, entry 11) was determined by means of X-ray diffraction studies, and the crystallographic data and structural refinement parameters have been deposited with the CCDC. Colorless crystals of **8k** were obtained by slow diffusion of hexane over a saturated ethyl acetate solution. The compound crystallized in the prism space group C2/c. The crystallographic structure and atom numbering are given in Figure 2, which shows a perspective view of compound **8k**. The crystallographic structure reveals the orientation of the three aryl substituents with respect to the pyrrolopyrimidine system.

CONCLUSIONS

The chemistry outlined here provides an efficient pathway for the synthesis of triarylated pyrrolo[2,3-d]pyrimidines through three cross-coupling reactions catalyzed by palladium under microwave irradiation. The high yields obtained in the different steps suggest that this process has considerable promise for the synthesis of pyrrolopyrimidines bearing three different (or equivalent) arylated substituents (overall yield 35–50%). This method allows arylation at C-4, C-5 and C-6 of pyrrolopyrimidines and should be of interest for the synthesis of medicinal and photoactive structures.

EXPERIMENTAL SECTION

General: Microwave-assisted reactions were carried out with a Biotage Initiator Microwave synthesis instrument and the internal temperature was measured by an IR sensor. The reactions were monitored by thin-layer chromatography (TLC) analysis using silica gel (60 F254) plates. Compounds were visualized by UV irradiation. Column chromatography was performed with silica gel 60 (230–400 mesh, 0.040–0.063 mm). Melting points (mp) were obtained with a melting point apparatus with a digital thermometer in open capillary tubes and are reported without correction. IR spectra were obtained with an FTIR Infrared Spectrophotometer. ¹H and ¹³C NMR spectra were recorded at 250 MHz (¹³C, 63 MHz), 300 (¹³C, 75.5 MHz), or 400 MHz (¹³C, 100 MHz). Chemical shifts (δ) (multiplicity, coupling constants and integration) are reported in parts per million (ppm) relative to the central peak of the solvent: CDCl₃ [δ = 7.26 (H) and 77.16 (C) ppm], CD₃OD [δ = 3.31 (H) and 49.45 (C) ppm], [D₆]DMSO [δ = 2.49 (H) and 39.51 (C) ppm] as internal standards. The following abbreviations are used for the proton spectra multiplicities: s singlet, d doublet, t triplet, q quadruplet, m multiplet, dd doublet of doublets, dt doublet of triplets and br. broad signal. Coupling constants (J) are reported in Hertz (Hz). High-resolution mass spectra (HRMS) were recorded with a time-of-flight mass spectrometer fitted with either an electrospray (ESI) or atmospheric pressure ionization (APCI). All reagents were of high quality or were purified before use. Organic solvents were of analytical grade or were purified by standard procedures.

6-Chloro-N-methylpyrimidin-4-amine (2): To a solution of 4,6-dichloropyrimidine (1; 200 mg, 1.34 mmol) in 2-propanol (1.3 mL), triethylamine (0.22 mL, 1.61 mmol) and MeNH₂·HCl (108.8 mg, 1.61 mmol) were added. The reaction was stirred and heated to the reflux temperature of 2-propanol until complete consumption of the starting material as determined by TLC analysis (4 h). After cooling, the solvent was removed under vacuo. Water (15 mL) was added and the mixture was extracted with ethyl acetate (3 × 15 mL). The combined organic layers were washed with NH₄Cl (10 mL). Evaporation of the solvent under reduced pressure gave the crude product, which was purified by silica gel column chromatography (CH₂Cl₂/ethyl acetate, 6:4) to afford 2 (150.7 mg, 1.05 mmol, 78%) as a white solid, m.p. 136–138 °C (EtOAc) [136–138 °C (hexane)]. [26] R_f = 0.22 (CH₂Cl₂/ethyl acetate, 6:4). IR (ATR diamond): $\tilde{\nu}$ = 3246, 3083, 2967, 1619, 1568, 1429, 1384, 1329, 1227, 1143, 1092, 976, 887, 834, 741, 708 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 2.92 (d, J = 4.3 Hz, 3 H), 5.87 (s, 1 H), 6.33 (s, 1 H), 8.31 (s, 1 H) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 28.3, 99.9, 158.2, 159.8, 164.0 ppm. HRMS (ESI): calcd. For C₅H₇ClN₃ [M + H]⁺ 144.0323; found 144.0322.

6-Chloro-5-iodo-N-methylpyrimidin-4-amine (3): A solution of pyrimidine 2 (2 g, 13.9 mmol) in acetonitrile (30 mL) was transferred to a microwave reaction tube and irradiated in a microwave oven twice at 100 °C for 15 min with previous addition of NIS (9.4 g, 41.7 mmol). The progress of the reaction was monitored by TLC (CH₂Cl₂/ethyl acetate, 6:4). After cooling, the solvent was removed, then CH₂Cl₂ (50 mL) was added and the organic layer was washed with aqueous saturated Na₂S₂O₃ (2 × 50 mL), and NaOH (10%, 2 × 50 mL). Evaporation of the solvent under reduced pressure gave the crude product, which was purified by recrystallization from ethanol (20 mL) to afford 3 (3.27 g, 12.1 mmol, 87%) as pale yellow crystals; m.p. 146–148 °C (EtOH). R_f = 0.50 (CH₂Cl₂/ethyl acetate, 6:4). IR (ATR diamond): $\tilde{\nu}$ = 3292, 2936, 1557, 1499, 1389, 1326, 1263, 1237, 1179, 1122, 1080, 1001, 888, 762 cm⁻¹. ¹H NMR (250 MHz, CDCl₃): δ = 3.06 (d, J = 4.9 Hz, 3 H), 5.64 (s, 1 H), 8.24 (s, 1 H) ppm. ¹³C NMR (63 MHz, CDCl₃): δ = 29.45, 79.6, 157.3, 162.2, 163.0 ppm. HRMS (ESI): calcd. for C₅H₆ClIN₃ [M + H]⁺ 269.9289; found 269.9291.

General Procedure A (Sonogashira Coupling): A mixture containing 3 (1.0 mmol), alkyne (2.0 mmol), Pd(dba)₂ (0.03 mmol), tri(2-furyl) phosphine (0.06 mmol), and CuI (0.04 mmol) in anhydrous THF (1 mL) and anhydrous triethylamine (3.5 mL) was transferred to a microwave reaction tube and irradiated in a microwave oven at 100 °C for 30 min. The progress of the reaction was monitored by TLC. After cooling, the mixture was diluted with aqueous saturated NH₄Cl (15 mL) and the aqueous

phase was extracted with ethyl acetate (3–20 mL). The combined organic phases were dried with MgSO₄ and filtered through Celite. Evaporation of the solvent under reduced pressure gave the crude product, which was purified by silica gel column chromatography.

6-Chloro-N-methyl-5-(2-phenylethynyl)pyrimidin-4-amine (4a); General Procedure A: The reaction was carried out by following general procedure A starting from the iodinated pyrimidine 3 (1 g, 3.71 mmol) and phenylacetylene (0.82 mL, 7.42 mmol). The crude product was purified by silica gel column chromatography (petroleum ether/CH₂Cl₂/acetone, 8:1:1) followed by recrystallization (CH₂Cl₂/pentane) to afford 4a (842.2 mg, 3.46 mmol, 93%) as a white solid; m.p. 113–115 °C (pentane). R_f = 0.28 (petroleum ether/CH₂Cl₂/acetone, 8:1:1). IR (ATR diamond): $\tilde{\nu}$ = 3397, 1564, 1490.1392, 1274, 1230, 1138, 1088, 906, 848, 749, 684 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 3.12 (d, J = 5.0 Hz, 3 H), 5.76 (s, 1 H), 7.34–7.44 (m, 3 H), 7.51–7.60 (m, 2 H), 8.35 (s, 1 H) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 28.6, 79.6, 101.0, 102.3, 122.0, 128.7, 129.5, 131.8, 156.2, 159.2, 163.0 ppm. HRMS (ESI): calcd. For C₁₃H₁₁ClN₃ [M + H]⁺ 244.0636; found 244.0637.

6-Chloro-5-[2-(4-methoxyphenyl)ethynyl]-N-methylpyrimidin-4-amine (4b): The reaction was carried out by following general procedure A starting from the iodinated pyrimidine 3 (150 mg, 0.56 mmol) and 1-ethynyl-4-methoxybenzene (147.1 mg, 1.11 mmol). The crude product was purified by silica gel column chromatography (petroleum ether/CH₂Cl₂/acetone, 70:15:15) followed by recrystallization (CH₂Cl₂/pentane) to afford 4b (143.8 mg, 0.53 mmol, 94%) as a white solid; m.p. 131–133 °C (pentane). R_f = 0.30 (petroleum ether/CH₂Cl₂/acetone, 70:15:15). IR (ATR diamond): $\tilde{\nu}$ = 3312, 2927, 2202, 1564, 1504, 1460, 1388, 1352, 1278, 1251, 1170, 1137.1086, 1031, 902, 831, 779, 731 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 3.11 (d, J = 5.0 Hz, 3 H), 3.84 (s, 3 H), 5.75 (s, 1 H), 6.90 (d, J = 8.7 Hz, 2 H), 7.48 (d, J = 8.7 Hz, 2 H), 8.33 (s, 1 H) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 28.5, 55.5, 78.3, 101.3, 102.5, 114.0, 114.3, 133.3, 156.0, 158.9, 160.6, 162.9 ppm. HRMS (ESI): calcd. for C₁₄H₁₃ClN₃O [M + H]⁺ 274.0742; found 274.0743.

6-Chloro-N-methyl-5-{2-[4-(trifluoromethyl)phenyl]ethynyl}-pyrimidin-4-amine (4c): The reaction was carried out by following general procedure A starting from the iodinated pyrimidine 3 (150 mg, 0.56 mmol) and 1-ethynyl-4-(trifluoromethylbenzene) (0.2 mL, 1.11 mmol). The crude product was purified by silica gel column chromatography (petroleum ether/CH₂Cl₂/acetone, 8:1:1) followed by recrystallization (CH₂Cl₂/pentane) to afford 4c (172.7 mg, 0.55 mmol, 99%) as a pale-yellow solid; m.p. 107–109 °C (pentane). R_f = 0.17 (petroleum ether/CH₂Cl₂/acetone, 8:1:1). IR (ATR diamond): $\tilde{\nu}$ = 3360, 2925, 1562.1509, 1396, 1322, 1276, 1228, 1167, 1090, 1064, 903, 836, 781, 736 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 3.13 (d, J = 5.0 Hz, 3 H), 5.73 (s, 1 H), 7.64 (s, 4 H), 8.37 (s, 1 H) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 28.6, 81.9, 100.3, 100.6, 123.9 (J = 272 Hz), 125.6 (J = 4 Hz), 125.8 (J = 1 Hz), 131.1 (J = 33 Hz), 132.0, 156.7, 159.8, 163.0 ppm. ¹⁹F NMR (376 MHz, CDCl₃): δ = –62.94 (CF₃) ppm. HRMS (ESI): calcd. for C₁₄H₁₀ClF₃N₃ [M + H]⁺ 312.0510; found 312.0511.

General Procedure B (Cyclization): A solution of alkyne 4a–c (1.0 mmol) and cesium carbonate (1.0 mmol) in acetonitrile (4.5 mL) was transferred to a microwave reaction tube and irradiated in a microwave oven at 100 °C for the required time. The progress of the reaction was monitored by TLC. After cooling, the solvent was removed under vacuo. Water (20 mL) was added to the mixture and the crude material was extracted with ethyl acetate (3–15 mL). The combined organic phases were washed with aqueous saturated Na₂CO₃ (15 mL) and brine (15 mL). Purification was performed by column chromatography on silica gel to give the corresponding pure product 5.

4-Chloro-7-methyl-6-phenyl-7H-pyrrolo[2,3-d]pyrimidine (5a): The reaction was carried out by following general procedure B starting from the alkyne 4a (800 mg, 3.28 mmol) and microwave irradiation was applied for 30 min. The crude product was purified by silica gel column chromatography (petroleum ether/CH₂Cl₂/acetone, 8:1:1) followed by recrystallization (CH₂Cl₂/pentane) to afford 5a (795.2 mg, 3.26 mmol, 99%) as a pale-yellow solid; m.p. 151–153 °C (pentane). R_f = 0.18 (petroleum ether/CH₂Cl₂/acetone, 8:1:1). IR (ATR diamond): $\tilde{\nu}$ = 3741, 2946, 1585, 1541, 1485, 1432, 1405, 1347, 1237, 1184, 1127, 1013, 938, 855, 755, 702 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 3.85 (s, 3 H), 6.62 (s, 1 H), 7.44–7.58 (m, 5 H), 8.64 (s, 1 H) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 30.5, 98.8,

117.7, 129.0, 129.3, 129.4, 130.9, 143.4, 150.6, 151.4, 152.7 ppm. HRMS (ESI): calcd. for $C_{13}H_{11}ClN_3$ $[M + H]^+$ 244.0636; found 244.0637.

4-Chloro-6-(4-methoxyphenyl)-7-methyl-7H-pyrrolo[2,3-d]pyrimidine (5b): The reaction was carried out by following general procedure B starting from the alkyne 4b (267.6 mg, 0.98 mmol) and microwave irradiation was applied for 90 min. The crude product was purified by silica gel column chromatography (petroleum ether/ CH_2Cl_2 /acetone, 8:1:1) followed by recrystallization (CH_2Cl_2 /pentane) to afford 5b (254.2 mg, 0.93 mmol, 95%) as a pale-yellow solid; m.p. 134–136 °C (pentane). R_f = 0.24 (petroleum ether/ CH_2Cl_2 /acetone, 8:1:1). IR (ATR diamond): $\tilde{\nu}$ = 3838, 3744, 2950, 1613, 1590, 1544, 1492, 1348, 1291, 1244, 1174, 1133, 1034, 940, 839, 805, 772, 741, 706 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$): δ = 3.83 (s, 3 H), 3.88 (s, 3 H), 6.56 (s, 1 H), 7.00–7.07 (m, 2 H), 7.43–7.50 (m, 2 H), 8.63 (s, 1 H) ppm. ^{13}C NMR (100 MHz, $CDCl_3$): δ = 30.5, 55.6, 98.1, 114.5, 117.8, 123.2, 130.7, 143.4, 150.3, 151.1, 152.7, 160.6 ppm. HRMS (ESI): calcd. for $C_{14}H_{13}ClN_3O$ $[M + H]^+$ 274.0742; found 274.0744.

4-Chloro-7-methyl-6-[4-(trifluoromethyl)phenyl]-7H-pyrrolo[2,3-d]-pyrimidine (5c): The reaction was carried out by following general procedure B starting from the alkyne 4c (296.6 mg, 0.95 mmol) and microwave irradiation was applied for 30 min. The crude product was purified by silica gel column chromatography (petroleum ether/ CH_2Cl_2 /acetone, 8:1:1) followed by recrystallization (CH_2Cl_2 /pentane) to afford 5c (98%, 290.6 mg, 0.93 mmol) as a yellow solid; m.p. 167–169 °C (pentane). R_f = 0.27 (petroleum ether/ CH_2Cl_2 /acetone, 8:1:1). IR (ATR diamond): $\tilde{\nu}$ = 3743, 2927, 1583, 1544, 1502, 1475, 1447, 1411, 1319, 1262, 1235, 1168, 1119, 1065, 1013, 944, 859, 804, 774, 745, 713 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$): δ = 3.87 (s, 3 H), 6.70 (s, 1 H), 7.68 (d, J = 8.2 Hz, 2 H), 7.79 (d, J = 8.2 Hz, 2 H), 8.68 (s, 1 H) ppm. ^{13}C NMR (100 MHz, $CDCl_3$): δ = 30.6, 99.9, 117.6, 124.0 (J = 272 Hz), 126.1 (J = 4 Hz), 129.6, 131.4 (J = 33 Hz), 134.5 (J = 1 Hz), 141.5, 152.0, 152.9 ppm. ^{19}F NMR (376 MHz, $CDCl_3$): δ = –62.80 (CF_3) ppm. HRMS (ESI): calcd. for $C_{14}H_{10}ClF_3N_3$ $[M + H]^+$ 312.0510; found 312.0513.

General Procedure C (Suzuki–Miyaura Coupling at C-4): Under argon, a mixture of compound 5a–c (1.0 mmol), boronic acid (1.05 mmol), sodium carbonate (2.0 mmol), and tetrakis(triphenylphosphine) palladium (0.02 mmol) in a degassed solvent mixture of DME (3.8 mL) and H_2O (0.6 mL) was transferred to a microwave reaction tube and irradiated in a microwave oven at 100 °C for 60 min. The progress of the reaction was monitored by TLC. After cooling, the mixture was diluted with a mixture of brine and water (1:1, 20 mL) and the aqueous solution was extracted with ethyl acetate (3–20 mL). Evaporation of the solvent under reduced pressure gave the crude product, which was purified by silica gel column chromatography.

7-Methyl-6-phenyl-4-(3-tolyl)-7H-pyrrolo[2,3-d]pyrimidine (6a): The reaction was carried out by following general procedure C starting from 4-chloropyrimidine 5a (450 mg, 1.85 mmol) and 3-tolylboronic acid (263.6 mg, 1.94 mmol). The crude product was purified by silica gel column chromatography (CH_2Cl_2 /ethyl acetate, 7:3) to afford 6a (507.4 mg, 1.69 mmol, 91%) as a pale-yellow solid. R_f = 0.32 (CH_2Cl_2 /ethyl acetate, 6:4); m.p. 109–111 °C (ethyl acetate). IR (ATR diamond): $\tilde{\nu}$ = 3836, 3741, 2922, 1698, 1560, 1490, 1468, 1396, 1343, 1268, 1217, 1078, 1014, 892, 814, 772, 745 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$): δ = 2.47 (s, 3 H), 3.89 (s, 3 H), 6.86 (s, 1 H), 7.30–7.32 (m, 1 H), 7.40–7.58 (m, 6 H), 7.91–8.01 (m, 2 H), 8.99 (s, 1 H) ppm. ^{13}C NMR (100 MHz, $CDCl_3$): δ = 21.5, 29.9, 99.8, 115.8, 125.9, 128.5, 128.7, 128.8, 129.1, 129.2, 130.6, 138.2, 138.4, 142.7, 151.3, 153.3, 156.8 ppm. HRMS (ESI): calcd. For $C_{20}H_{18}N_3$ $[M + H]^+$ 300.1495; found 300.1498.

7-Methyl-6-phenyl-4-(2-tolyl)-7H-pyrrolo[2,3-d]pyrimidine (6b): The reaction was carried out by following general procedure C starting from 4-chloropyrimidine 5a (150 mg, 0.62 mmol) and 2-tolylboronic acid (87.9 mg, 0.65 mmol). The crude product was purified by silica gel column chromatography (CH_2Cl_2 /ethyl acetate, 8:2) to afford 6b (181.7 mg, 0.61 mmol, 99%) as a white solid. R_f = 0.45 (CH_2Cl_2 /ethyl acetate, 8:2); m.p. 104–106 °C (ethyl acetate). IR (ATR diamond): $\tilde{\nu}$ = 2919,

2849, 1738, 1567, 1541, 1491, 1443, 1404, 1370, 1340, 1320, 1261, 1218, 1135, 1011, 935, 865, 787, 757, 748, 727, 699 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 2.39 (s, 3 H), 3.91 (s, 3 H), 6.47 (s, 1 H), 7.27–7.40 (m, 3 H), 7.44–7.56 (m, 6 H), 8.99 (s, 1 H) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 20.3, 30.1, 99.9, 117.8, 125.8, 128.9, 129.0, 129.1, 129.2, 129.8, 131.1, 131.4, 136.6, 137.5, 142.8, 151.1, 152.9, 159.3 ppm. HRMS (ESI): calcd. for C₂₀H₁₈N₃ [M + H]⁺ 300.1495; found 300.1498.

7-Methyl-6-phenyl-4-(4-tolyl)-7H-pyrrolo[2,3-d]pyrimidine (6c): The reaction was carried out by following general procedure C starting from 4-chloropyrimidine 5a (150 mg, 0.62 mmol) and 4-tolylboronic acid (87.9 mg, 0.65 mmol). The crude product was purified by silica gel column chromatography (CH₂Cl₂/ethyl acetate, 8:2) to afford 6c (178.4 mg, 0.60 mmol, 96%) as a white solid. R_f = 0.48 (CH₂Cl₂/ethyl acetate, 8:2); m.p. 94–96 °C (ethyl acetate). IR (ATR diamond): ν̃ = 3052, 2921, 2855, 1732, 1609, 1554, 1443, 1331, 1261, 1180, 1017, 938, 834, 785, 759, 698 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 2.42 (s, 3 H), 3.85 (s, 3 H), 6.84 (s, 1 H), 7.32 (d, J = 8.0 Hz, 2 H), 7.48–7.51 (m, 5 H), 8.06 (d, J = 8.0 Hz, 2 H), 8.96 (s, 1 H) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 21.6, 30.2, 100.1, 115.8, 128.9, 128.9, 129.0, 129.3, 129.6, 131.5, 135.6, 140.3, 142.8, 151.4, 153.5, 156.8 ppm. HRMS (ESI): calcd. For C₂₀H₁₈N₃ [M + H]⁺ 300.1495; found 300.1500.

4-(4-Methoxyphenyl)-7-methyl-6-phenyl-7H-pyrrolo[2,3-d]pyrimidine (6d): The reaction was carried out by following general procedure C starting from 4-chloropyrimidine 5a (150 mg, 0.62 mmol) and 4-methoxyphenylboronic acid (98.2 mg, 0.65 mmol). The crude product was purified by silica gel column chromatography (CH₂Cl₂/ethyl acetate, 6:4) to afford 6d (194.5 mg, 0.62 mmol, 100%) as a yellow solid. R_f = 0.48 (CH₂Cl₂/ethyl acetate, 8:2); m.p. 174–176 °C (ethyl acetate). IR (ATR diamond): ν̃ = 3054, 2921, 2845, 1550, 1512, 1439, 1328, 1247, 1172, 1021, 937, 851, 756 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 3.89 (s, 3 H), 3.90 (s, 3 H), 6.87 (s, 1 H), 7.07 (d, J = 8.2 Hz, 2 H), 7.47–7.59 (m, 5 H), 8.17 (d, J = 8.2 Hz, 2 H), 8.96 (s, 1 H) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 30.1, 55.8, 100.0, 114.3, 115.4, 128.9, 129.0, 129.3, 130.4, 131.0, 131.6, 142.6, 151.4, 153.4, 160.6, 161.3 ppm. HRMS (ESI): calcd. for C₂₀H₁₈N₃O [M + H]⁺ 316.1444; found 316.1447.

7-Methyl-6-phenyl-4-[4-(trifluoromethyl)phenyl]-7H-pyrrolo[2,3-d]-pyrimidine (6e): The reaction was carried out by following general procedure C starting from 4-chloropyrimidine 5a (150 mg, 0.62 mmol) and 4-(trifluoromethyl)phenylboronic acid (122.8 mg, 0.65 mmol). The crude product was purified by silica gel column chromatography (CH₂Cl₂/ethyl acetate, 8:2) to afford 6e (213.7 mg, 0.60 mmol, 98%) as a beige solid. R_f = 0.57 (CH₂Cl₂/ethyl acetate, 8:2); m.p. 137–139 °C (ethyl acetate). IR (ATR diamond): ν̃ = 2960, 2920, 2849, 1736, 1620, 1555, 1493, 1466, 1400, 1327, 1307, 1265, 1219, 1185, 1165, 1103, 1064, 1014, 936, 850, 808, 783, 753, 730, 699 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 3.92 (s, 3 H), 6.85 (s, 1 H), 7.49–7.57 (m, 5 H), 7.81 (d, J = 8.0 Hz, 2 H), 8.28 (d, J = 8.0 Hz, 2 H), 9.02 (s, 1 H) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 30.3, 99.4, 116.2, 124.2 (J = 271 Hz), 125.8 (J = 4 Hz), 129.0, 129.2, 129.3, 129.4, 131.2, 131.7 (J = 32 Hz), 141.8, 143.8, 151.4, 153.7, 154.9 ppm. ¹⁹F NMR (376 MHz, CDCl₃): δ = -62.68 (CF₃) ppm. HRMS (ESI): calcd. for C₂₀H₁₅N₃F₃ [M + H]⁺ 354.1213; found 354.1214.

6-(4-Methoxyphenyl)-7-methyl-4-(3-tolyl)-7H-pyrrolo[2,3-d]pyrimidine (6f): The reaction was carried out by following general procedure C starting from 4-chloropyrimidine 5b (150 mg, 0.55 mmol) and 3-tolylboronic acid (78.2 mg, 0.58 mmol). The crude product was purified by silica gel column chromatography (CH₂Cl₂/ethyl acetate, 8:2) to afford 6f (173.9 mg, 0.53 mmol, 96%) as an offwhite solid; m.p. 95–97 °C (ethyl acetate). R_f = 0.38 (CH₂Cl₂/ethyl acetate, 8:2). IR (ATR diamond): ν̃ = 2921, 2855, 2153, 1968, 1615, 1558, 1497, 1442, 1339, 1295, 1252, 1173, 1013, 893, 826, 786, 763, 700 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 2.59 (s, 3 H), 3.97 (s, 3 H), 3.98 (s, 3 H), 6.92 (s, 1 H), 7.14 (d, J = 8.8 Hz, 2 H), 7.42 (d, J = 7.5 Hz, 1 H), 7.54 (t, J = 7.5 Hz, 1 H), 7.59 (d, J = 8.8 Hz, 2 H), 8.04–8.14 (m, 2 H), 9.11 (s, 1 H) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 21.7, 30.1, 55.5, 99.3, 114.4, 116.0, 123.8, 126.1, 128.7, 129.4, 130.6, 130.8, 138.4, 138.9, 142.9, 151.2, 153.3, 156.6, 160.3 ppm. HRMS (ESI): calcd. for C₂₁H₂₀N₃O [M + H]⁺ 330.1601; found 330.1602.

7-Methyl-4-(3-tolyl)-6-[4-(trifluoromethyl)phenyl]-7H-pyrrolo[2,3-d]-pyrimidine (6g): The reaction was carried out by following general procedure C starting from 4-chloropyrimidine 5c (150 mg, 0.48

mmol) and 3-tolylboronic acid (68.7 mg, 0.51 mmol). The crude product was purified by silica gel column chromatography (CH₂Cl₂/ethyl acetate, 8:2) to afford 6g (169.4 mg, 0.46 mmol, 96%) as an off-white solid. R_f = 0.43 (CH₂Cl₂/ethyl acetate, 8:2); m.p. 109–111 °C (ethyl acetate). IR (ATR diamond): $\tilde{\nu}$ = 2921, 2849, 1615, 1551, 1440, 1322, 1160, 1114, 1068, 894, 847, 767, 702 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 2.45 (s, 3 H), 3.87 (s, 3 H), 6.89 (s, 1 H), 7.30 (d, J = 7.6 Hz, 1 H), 7.41 (t, J = 7.6 Hz, 1 H), 7.67 (d, J = 8.2 Hz, 2 H), 7.75 (d, J = 8.2 Hz, 2 H), 7.91 (d, J = 7.6 Hz, 1 H), 7.95 (s, 1 H), 8.99 (s, 1 H) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 21.7, 30.3, 101.2, 115.8, 124.0 (J = 271 Hz), 125.9 (J = 4 Hz), 126.1, 128.8, 129.5, 129.6, 131.0 (J = 32 Hz), 131.1, 135.1, 138.1, 138.8, 141.1, 151.9, 153.6, 157.6 ppm. ¹⁹F NMR (376 MHz, CDCl₃): δ = -62.73 (CF₃) ppm. HRMS (ESI): calcd. for C₂₁H₁₇N₃F₃ [M + H]⁺ 368.1369; found 368.1371.

General Procedure D (Aryl Iodination): A solution of 6a–g (1.0 mmol) and NIS (1.1 mmol) in acetonitrile (4.5 mL) was transferred to a microwave reaction tube and irradiated in a microwave oven at 100 °C for 30 min. The progress of the reaction was monitored by TLC. After cooling, the solvent was removed under vacuo. Dichloromethane (20 mL) was added and the organic layer was washed with aqueous saturated Na₂S₂O₃ (2–16 mL), and NaOH (10%, 2–16 mL). Evaporation of the solvent under reduced pressure gave the crude product, which did not require further purification.

5-Iodo-7-methyl-6-phenyl-4-(3-tolyl)-7H-pyrrolo[2,3-d]pyrimidine (7a): The reaction was carried out by following general procedure D starting from diaryl pyrimidine 6a (500 mg, 1.67 mmol) to afford 7a (710.0 mg, 1.67 mmol, 100 %) as a yellow solid. R_f = 0.27 (CH₂Cl₂/ethyl acetate, 8:2); m.p. 188–190 °C (ethyl acetate). IR (ATR diamond): $\tilde{\nu}$ = 3045, 2913, 1547, 1439, 1402, 1330, 1254, 1182, 1088, 1020, 958, 915, 783, 764, 708, 682 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 2.45 (s, 3 H), 3.77 (s, 3 H), 7.32 (d, J = 7.6 Hz, 1 H), 7.40 (t, J = 7.6 Hz, 1 H), 7.43–7.47 (m, 2 H), 7.49–7.61 (m, 5 H), 8.98 (s, 1 H) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 21.6, 30.9, 55.2, 117.3, 127.7, 128.2, 128.8, 129.7, 130.3, 130.9, 131.3, 131.8, 135.8, 137.3, 144.5, 151.1, 152.5, 160.5 ppm. HRMS (ESI): calcd. for C₂₀H₁₇N₃ [M + H]⁺ 426.0462; found 426.0463.

5-Iodo-7-methyl-6-phenyl-4-(2-tolyl)-7H-pyrrolo[2,3-d]pyrimidine (7b): The reaction was carried out by following general procedure D starting from diaryl pyrimidine 6b (150 mg, 0.5 mmol) to afford 7b (211.5 mg, 0.50 mmol, 99 %) as a yellow solid. R_f = 0.51 (CH₂Cl₂/ethyl acetate, 8:2); m.p. 185–187 °C (ethyl acetate). IR (ATR diamond): $\tilde{\nu}$ = 2917, 2849, 1736, 1552, 1438, 1403, 1334, 1242, 1176, 1120, 1089, 951, 887, 796, 765, 725, 701 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 2.16 (s, 3 H), 3.79 (s, 3 H), 7.30–7.32 (m, 3 H), 7.37–7.41 (m, 1 H), 7.43–7.46 (m, 2 H), 7.50–7.54 (m, 3 H), 8.98 (s, 1 H) ppm. ¹³C NMR (63 MHz, CDCl₃): δ = 20.3, 30.9, 54.9, 118.2, 125.3, 128.8, 129.7, 129.9, 130.0, 130.8, 130.9, 135.7, 136.6, 142.8, 136.9, 151.3, 152.0, 162.2 ppm. HRMS (ESI): calcd. for C₂₀H₁₇N₃ [M + H]⁺ 426.0462; found 426.0463.

5-Iodo-7-methyl-6-phenyl-4-(4-tolyl)-7H-pyrrolo[2,3-d]pyrimidine (7c): The reaction was carried out by following general procedure D starting from diaryl pyrimidine 6c (150 mg, 0.50 mmol) to afford 7c (212.3 mg, 0.50 mmol, 100%) as an off-white solid. R_f = 0.29 (CH₂Cl₂/ethyl acetate, 8:2); m.p. 204–206 °C (ethyl acetate). IR (ATR diamond): $\tilde{\nu}$ = 2960, 2923, 2849, 1612, 1557, 1513, 1480, 1459, 1441, 1335, 1287, 1243, 1224, 1177, 1033, 953, 827, 764, 703 cm⁻¹. ¹H NMR (250 MHz, CDCl₃): δ = 2.45 (s, 3 H), 3.77 (s, 3 H), 7.32 (d, J = 7.8 Hz, 2 H), 7.39–7.49 (m, 2 H), 7.52–7.55 (m, 3 H), 7.65 (d, J = 7.8 Hz, 1 H), 8.97 (s, 1 H) ppm. ¹³C NMR (63 MHz, CDCl₃): δ = 21.7, 31.0, 55.2, 117.3, 128.5, 128.8, 129.7, 130.9, 131.1, 131.3, 133.2, 139.6, 144.4, 151.1, 152.5, 160.5 ppm. HRMS (ESI): calcd. for C₂₀H₁₇N₃ [M + H]⁺ 426.0462; found 426.0463.

5-Iodo-4-(4-methoxyphenyl)-7-methyl-6-phenyl-7H-pyrrolo[2,3-d]pyrimidine (7d): The reaction was carried out by following general procedure D starting from diaryl pyrimidine 6d (150 mg, 0.48 mmol) to afford 7d (209.1 mg, 0.48 mmol, 100%) as a yellow solid. R_f = 0.51 (CH₂Cl₂/ethyl acetate, 8:2); m.p. 196–198 °C (ethyl acetate). IR (ATR diamond): $\tilde{\nu}$ = 2921, 2849, 1608, 1556, 1513, 1439, 1321, 1293, 1243, 1174, 1025, 951, 885, 833, 797, 764, 739, 704 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 3.77 (s, 3 H), 3.89 (s, 3 H), 7.04 (d, J = 8.8 Hz, 2 H), 7.41–7.49 (m, 2 H), 7.50–7.58 (m, 3 H), 7.73 (d, J = 8.8 Hz, 2 H), 8.96 (s, 1 H) ppm. ¹³C NMR (63 MHz, CDCl₃): δ = 31.0, 55.4, 55.5, 113.2, 117.2,

128.5, 128.8, 129.7, 130.9, 131.3, 132.7, 144.4, 151.1, 152.6, 160.0, 161.0 ppm. HRMS (ESI): calcd. for C₂₀H₁₇N₃O [M + H]⁺ 442.0411; found 442.0412.

5-Iodo-7-methyl-6-phenyl-4-[4-(trifluoromethyl)phenyl]-7H-pyrrolo-[2,3-d]pyrimidine (7e): The reaction was carried out by following general procedure D starting from diaryl pyrimidine 6e (150 mg, 0.42 mmol) to afford 7e (202.4 mg, 0.42 mmol, 99%) as a beige solid. R_f = 0.60 (CH₂Cl₂/ethyl acetate, 8:2); m.p. 198–200 °C (ethyl acetate). IR (ATR diamond): $\tilde{\nu}$ = 2922, 2852, 1733, 1557, 1476, 1440, 1400, 1325, 1244, 1155, 1117, 1107, 1062, 1017, 948, 885, 843, 799, 762.704 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 3.79 (s, 3 H), 7.41–7.48 (m, 2 H), 7.51–7.59 (m, 3 H), 7.78 (d, J = 8.4 Hz, 2 H), 7.86 (d, J = 8.4 Hz, 2 H), 8.99 (s, 1 H) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 30.9, 54.4, 117.4, 124.2 (J = 272 Hz), 124.6 (J = 4 Hz), 128.8, 129.8, 130.7, 130.8, 131.3, 131.4 (J = 32 Hz), 139.4 (J = 1 Hz), 145.0, 151.0, 152.5, 158.5 ppm. ¹⁹F NMR (376 MHz, CDCl₃): δ = -62.52 (CF₃) ppm. HRMS (ESI): calcd. for C₂₀H₁₄F₃N₃ [M + H]⁺ 480.0179; found 480.0180.

5-Iodo-6-(4-methoxyphenyl)-7-methyl-4-(3-tolyl)-7H-pyrrolo[2,3-d]pyrimidine (7f): The reaction was carried out by following general procedure D starting from diaryl pyrimidine 6f (150 mg, 0.46 mmol) to afford 7f (100%, 207.3 mg, 0.46 mmol) as a yellow solid. R_f = 0.32 (CH₂Cl₂/acetone, 9:1); m.p. 179–181 °C (acetone). IR (ATR diamond): $\tilde{\nu}$ = 2955, 2918, 2849, 1736, 1609, 1559, 1533, 1467, 1446, 1414, 1289, 1243, 1173, 1021, 954, 837, 782, 711 cm⁻¹. ¹H NMR (250 MHz, CDCl₃): δ = 2.45 (s, 3 H), 3.77 (s, 3 H), 3.90 (s, 3 H), 7.05 (d, J = 8.8 Hz, 2 H), 7.28–7.34 (m, 1 H), 7.35–7.43 (m, 3 H), 7.52–7.55 (m, 2 H), 8.96 (s, 1 H) ppm. ¹³C NMR (63 MHz, CDCl₃): δ = 21.6, 30.9, 55.2, 55.5, 114.2, 117.3, 123.2, 127.7, 128.2, 130.2, 131.8, 132.3, 135.9, 137.3, 144.5, 150.9, 152.5, 160.2, 160.6 ppm. HRMS (ESI): calcd. for C₂₁H₁₉N₃O [M + H]⁺ 456.0567; found 456.0566.

5-Iodo-7-methyl-4-(3-tolyl)-6-[4-(trifluoromethyl)phenyl]-7H-pyrrolo[2,3-d]pyrimidine (7g): The reaction was carried out by following general procedure D starting from diaryl pyrimidine 6g (150 mg, 0.41 mmol) to afford 7g (201.2 mg, 0.41 mmol, 100%) as a beige solid. R_f = 0.53 (CH₂Cl₂/ethyl acetate, 7:3); m.p. 137–139 °C (ethyl acetate). IR (ATR diamond): $\tilde{\nu}$ = 2923, 2855, 1736, 1555, 1447, 1407, 1320, 1251, 1163, 1127, 1067, 1017, 964, 911, 855, 791, 709 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 2.46 (s, 3 H), 3.79 (s, 3 H), 7.31–7.36 (m, 1 H), 7.38–7.45 (m, 1 H), 7.51–7.57 (m, 2 H), 7.61 (d, J = 8.7 Hz, 2 H), 7.82 (d, J = 8.7 Hz, 2 H), 9.01 (s, 1 H) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 21.5, 30.9, 55.8, 117.1, 123.8 (J = 274 Hz), 125.7 (J = 4 Hz), 127.7, 128.0, 130.4, 131.4, 131.6, 131.7 (J = 33 Hz), 134.8 (J = 1 Hz), 135.3, 137.3, 142.7, 151.2, 152.5, 160.8 ppm. ¹⁹F NMR (376 MHz, CDCl₃): δ = -62.82 (CF₃) ppm. HRMS (ESI): calcd. for C₂₁H₁₆F₃N₃ [M + H]⁺ 494.0336; found 494.0335.

General Procedure E (Suzuki–Miyaura Coupling at C-5): Under argon, a mixture of compound 7a–g (1.0 mmol), boronic acid (1.05 mmol), sodium carbonate (2.0 mmol), and dichlorobis(triphenylphosphine) palladium (0.05 mmol) in a degassed solvent mixture of DME (3.8 mL) and H₂O (0.6 mL) was transferred to a microwave reaction tube and irradiated in a microwave oven at 100 °C for 60 min. The progress of the reaction was monitored by TLC. After cooling, the mixture was diluted with a mixture of brine and water (1:1, 20 mL), and the aqueous solution was extracted with ethyl acetate (3 × 20 mL). Evaporation of the solvent under reduced pressure gave the crude product. A solution of the previous residue and NIS (1.1 mmol) in acetonitrile (4.5 mL) was transferred to a microwave reaction tube and irradiated in a microwave oven at 100 °C for 30 min. The progress of the reaction was monitored by TLC. After cooling, the solvent was removed, then dichloromethane (20 mL) was added and the organic layer was washed with aqueous saturated Na₂S₂O₃ (2 × 16 mL) and NaOH (10%, 2 × 16 mL). Evaporation of the solvent under reduced pressure gave the crude product, which was purified by silica gel column chromatography.

5-(4-Methoxyphenyl)-7-methyl-6-phenyl-4-(3-tolyl)-7H-pyrrolo-[2,3-d]pyrimidine (8a): The reaction was carried out by following general procedure E starting from the iodo derivative 7a (100 mg, 0.24 mmol) and 4-methoxyphenylboronic acid (39.3 mg, 0.26 mmol). The crude product was purified by silica gel column chromatography (CH₂Cl₂/acetone, 95:5) to afford 8a (83.9 mg, 0.21 mmol, 88%) as a white solid (8% starting material was also recovered). R_f = 0.11 (CH₂Cl₂/acetone, 95:5); m.p. 144–

146°C (acetone). IR (ATR diamond): $\tilde{\nu}$ = 2954, 1614, 1552, 1535, 1514, 1466, 1438, 1413, 1349, 1322, 1287, 1238, 1176, 1134, 1108, 1036, 964, 919, 847, 816, 790, 762, 746 cm^{-1} . ^1H NMR (400 MHz, $[\text{D}_6]\text{-DMSO}$): δ = 1.96 (s, 3 H), 3.64 (s, 3 H), 3.73 (s, 3 H), 6.52 (d, J = 8.8 Hz, 2 H), 6.68 (d, J = 8.8 Hz, 2 H), 6.88 (s, 1 H), 7.03–7.09 (m, 2 H), 7.21–7.27 (m, 1 H), 7.35–7.39 (m, 2 H), 7.39–7.43 (m, 3 H), 8.91 (s, 1 H) ppm. ^{13}C NMR (100 MHz, $[\text{D}_6]\text{DMSO}$): δ = 20.5, 29.6, 55.0, 112.9, 113.0, 114.2, 126.1, 126.2, 127.7, 128.3, 128.7, 129.1, 130.1, 130.8, 130.9, 131.6, 136.2, 137.0, 138.9, 150.6, 151.4, 157.6, 158.3 ppm. HRMS (ESI): calcd. for $\text{C}_{27}\text{H}_{24}\text{N}_3\text{O}$ $[\text{M} + \text{H}]^+$ 406.1914; found 406.1915.

7-Methyl-6-phenyl-4-(3-tolyl)-5-[4-(trifluoromethyl)phenyl]-7H-pyrrolo[2,3-d]pyrimidine (8b):

The reaction was carried out by following general procedure E starting from the iodo derivative 7a (100 mg, 0.24 mmol) and 4-trifluoromethylphenylboronic acid (49.1 mg, 0.26 mmol). The crude product was purified by silica gel column chromatography ($\text{CH}_2\text{Cl}_2/\text{acetone}$, 97.5:2.5 to 9:1) to afford 8b (69.8 mg, 0.16 mmol, 67%) as a beige solid (29% of starting material was also recovered). R_f = 0.22 ($\text{CH}_2\text{Cl}_2/\text{acetone}$, 9:1); m.p. 199–201 °C (acetone). IR (ATR diamond): $\tilde{\nu}$ = 3043, 2921, 2847, 1618, 1556, 1538, 1438, 1409, 1348, 1323, 1238, 1161, 1115, 1106, 1064, 1023, 962, 919, 856, 825, 789, 760, 711, 697 cm^{-1} . ^1H NMR (400 MHz, CDCl_3): δ = 1.98 (s, 3 H), 3.84 (s, 3 H), 6.85 (d, J = 8.0 Hz, 2 H), 6.90 (s, 1 H), 7.04 (d, J = 6.5 Hz, 2 H), 7.16 (d, J = 8.0 Hz, 2 H), 7.22 (d, J = 6.5 Hz, 1 H), 7.24–7.28 (m, 2 H), 7.39–7.41 (m, 3 H), 9.04 (s, 1 H) ppm. ^{13}C NMR (100 MHz, CDCl_3): δ = 20.9, 30.1, 113.2, 114.7, 124.2 (J = 4 Hz), 124.4 (J = 273 Hz), 126.4, 127.8, 128.1 (J = 32 Hz), 128.4, 128.8, 129.2, 129.8, 130.0, 130.9, 131.0, 131.1, 137.3, 138.1, 139.8, 151.5, 152.4, 159.6 ppm. ^{19}F NMR (376 MHz, CDCl_3): δ = –62.55 (CF_3) ppm. HRMS (ESI): calcd. for $\text{C}_{27}\text{H}_{21}\text{F}_3\text{N}_3$ $[\text{M} + \text{H}]^+$ 444.1682; found 444.1683.

7-Methyl-6-phenyl-4-(3-tolyl)-5-(4-tolyl)-7H-pyrrolo[2,3-d]pyrimidine (8c):

The reaction was carried out by following general procedure E starting from the iodo derivative 7a (100 mg, 0.24 mmol) and p-tolylboronic acid (35.2 mg, 0.26 mmol). The crude product was purified by silica gel column chromatography ($\text{CH}_2\text{Cl}_2/\text{acetone}$, 98:2 to 9:1) to afford 8c (64.8 mg, 0.17 mmol, 71%) as a darkbeige solid (26% of starting material was also recovered). R_f = 0.23 ($\text{CH}_2\text{Cl}_2/\text{acetone}$, 9:1); m.p. 204–206 °C (acetone). IR (ATR diamond): $\tilde{\nu}$ = 3034, 2917, 2849, 1737, 1553, 1537, 1516, 1436, 1413, 1348, 1320, 1273, 1238, 1210, 1177, 1130, 1026, 962, 918, 892, 845, 816, 788, 763, 743, 710, 698 cm^{-1} . ^1H NMR (300 MHz, CDCl_3): δ = 2.00 (s, 3 H), 2.21 (s, 3 H), 3.82 (s, 3 H), 6.64 (d, J = 8.0 Hz, 2 H), 6.73 (d, J = 8.0 Hz, 2 H), 6.92–7.08 (m, 3 H), 7.24–7.31 (m, 3 H), 7.33–7.40 (m, 3 H), 9.01 (s, 1 H) ppm. ^{13}C NMR (75.5 MHz, CDCl_3): δ = 21.2, 21.4, 30.3, 114.8, 115.2, 126.8, 127.7, 128.3, 128.7, 128.8, 129.6, 130.9, 131.0, 131.2, 131.3, 131.4, 135.8, 137.1, 137.5, 139.2, 151.2, 152.5, 159.7 ppm. HRMS (ESI): calcd. For $\text{C}_{27}\text{H}_{24}\text{N}_3$ $[\text{M} + \text{H}]^+$ 390.1965; found 390.1964.

7-Methyl-6-phenyl-4,5-bis-(3-tolyl)-7H-pyrrolo[2,3-d]pyrimidine (8d):

The reaction was carried out by following general procedure E starting from the iodo derivative 7a (100 mg, 0.24 mmol) and m-tolylboronic acid (35.2 mg, 0.26 mmol). The crude product was purified by silica gel column chromatography ($\text{CH}_2\text{Cl}_2/\text{acetone}$, 98:2 to 9:1) to afford 8d (69.6 mg, 0.18 mmol, 76%) as a white solid (22% of starting material was also recovered). R_f = 0.09 ($\text{CH}_2\text{Cl}_2/\text{acetone}$, 9:1); m.p. 151–153 °C (acetone). IR (ATR diamond): $\tilde{\nu}$ = 3028, 2919, 2847, 1557, 1538, 1492, 1463, 1441, 1418, 1349, 1324, 1241, 1193, 1166, 1128, 1086, 1025, 1000, 969, 924, 901, 870, 839, 812, 788, 763, 748, 719, 703 cm^{-1} . ^1H NMR (300 MHz, CDCl_3): δ = 1.93 (s, 3 H), 2.03 (s, 3 H), 3.82 (s, 3 H), 6.50 (s, 1 H), 6.60 (d, J = 7.1 Hz, 1 H), 6.78–6.86 (m, 2 H), 7.01–7.04 (m, 3 H), 7.21–7.23 (m, 1 H), 7.26–7.32 (m, 2 H), 7.34–7.39 (m, 3 H), 9.02 (s, 1 H) ppm. ^{13}C NMR (75.5 MHz, CDCl_3): δ = 21.1, 30.0, 114.7, 115.0, 126.5, 126.8, 127.3, 127.4, 127.7, 128.5, 128.7, 129.4, 130.7, 130.9, 131.1, 132.3, 133.7, 136.9, 137.6, 139.0, 151.1, 152.3, 159.7 ppm. HRMS (ESI): calcd. for $\text{C}_{27}\text{H}_{24}\text{N}_3$ $[\text{M} + \text{H}]^+$ 390.1965; found 390.1967.

7-Methyl-6-phenyl-4-(3-tolyl)-5-(2-tolyl)-7H-pyrrolo[2,3-d]pyrimidine (8e):

The reaction was carried out by following general procedure E starting from the iodo derivative 7a (100 mg, 0.24 mmol, 1.0 equiv.) and p-tolylboronic acid (35.2 mg, 0.26 mmol, 1.1 equiv.). The product was purified by chromatography on silica gel ($\text{CH}_2\text{Cl}_2/\text{acetone}$, 98:2 to 9:1) to afford starting material (82 %). Product 8e was not observed.

5-(4-Methoxyphenyl)-7-methyl-6-phenyl-4-(2-tolyl)-7H-pyrrolo-[2,3-d]pyrimidine (8f): The reaction was carried out by following general procedure E starting from the iodo derivative 7b (100 mg, 0.24 mmol) and 4-methoxyphenylboronic acid (39.3 mg, 0.26 mmol). The crude product was purified by silica gel column chromatography (CH₂Cl₂/acetone, 98:2 to 9:1) to afford 8f (89.5 mg, 0.22 mmol, 94%) as a beige solid (5% of starting material was also recovered). R_f = 0.42 (CH₂Cl₂/acetone, 9:1); m.p. 186–188 °C (acetone). IR (ATR diamond): $\tilde{\nu}$ = 3060, 2997, 2931, 1609, 1559, 1513, 1415, 1349, 1319, 1287, 1243, 1224, 1179, 1121, 1033, 953, 894, 837, 809, 774, 763, 729, 701 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 1.99 (s, 3 H), 3.65 (s, 3 H), 3.83 (s, 3 H), 6.34 (d, J = 8.6 Hz, 2 H), 6.55 (d, J = 8.6 Hz, 2 H), 6.94 (t, J = 7.4 Hz, 2 H), 7.01 (d, J = 7.4 Hz, 1 H), 7.09 (t, J = 7.4 Hz, 1 H), 7.26–7.28 (m, 2 H), 7.34–7.36 (m, 3 H), 9.00 (s, 1 H) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 19.8, 30.0, 55.2, 112.7, 114.4, 116.3, 125.1, 125.2, 128.3, 128.6, 128.7, 129.4, 129.7, 130.6, 131.1, 131.5, 135.8, 137.7, 138.9, 151.1, 151.8, 157.7, 160.1 ppm. HRMS (ESI): calcd. For C₂₇H₂₄N₃O [M + H]⁺ 406.1914; found 406.1913.

5-(4-Methoxyphenyl)-7-methyl-6-phenyl-4-(4-tolyl)-7H-pyrrolo-[2,3-d]pyrimidine (8g): The reaction was carried out by following general procedure E starting from the iodo derivative 7c (100 mg, 0.24 mmol) and 4-methoxyphenylboronic acid (39.3 mg, 0.26 mmol). The crude product was purified by silica gel column chromatography (CH₂Cl₂/acetone, 98:2 to 9:1) to afford 8g (61.9 mg, 0.15 mmol, 65%) as a beige solid (29% of starting material was also recovered). R_f = 0.21 (CH₂Cl₂/acetone, 8:2); m.p. 157–159 °C (acetone). IR (ATR diamond): $\tilde{\nu}$ = 2917, 2849, 1550, 1532, 1510, 1462, 1440, 1418, 1346, 1321, 1290, 1237, 1174, 1132, 1110, 1035, 953, 892, 831, 800, 783, 767, 713, 704 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 2.27 (s, 3 H), 3.69 (s, 3 H), 3.81 (s, 3 H), 6.46 (d, J = 8.7 Hz, 2 H), 6.64 (d, J = 8.7 Hz, 2 H), 6.87 (d, J = 8.0 Hz, 2 H), 7.19 (d, J = 8.0 Hz, 2 H), 7.25–7.29 (m, 2 H), 7.32–7.45 (m, 3 H), 9.00 (s, 1 H) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 21.4, 29.9, 55.3, 113.0, 114.2, 115.0, 126.4, 128.0, 128.5, 128.6, 129.6, 130.7, 131.1, 131.9, 134.7, 138.5, 138.9, 151.0, 152.2, 158.0, 159.4 ppm. HRMS (ESI): calcd. for C₂₇H₂₄N₃O [M + H]⁺ 406.1914; found 406.1916.

4,5-Bis(4-methoxyphenyl)-7-methyl-6-phenyl-7H-pyrrolo[2,3-d]pyrimidine (8h): The reaction was carried out by following general procedure E starting from the iodo derivative 7d (100 mg, 0.23 mmol) and 4-methoxyphenylboronic acid (37.9 mg, 0.25 mmol). The crude product was purified by silica gel column chromatography (CH₂Cl₂/acetone, 98:2 to 9:1) to afford 8h (87.7 mg, 0.21 mmol, 92%) as an off-white solid (6% of starting material was also recovered). R_f = 0.39 (CH₂Cl₂/acetone, 8:2); m.p. 172–174 °C (acetone). IR (ATR diamond): $\tilde{\nu}$ = 3039, 2923, 2832, 1605, 1552, 1532, 1510, 1455, 1435, 1416, 1347, 1322, 1301, 1289, 1246, 1235, 1173, 1130, 1108, 1034, 954, 927, 892, 842, 804, 783, 768, 704 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 3.71 (s, 3 H), 3.74 (s, 3 H), 3.81 (s, 3 H), 6.50 (d, J = 8.8 Hz, 2 H), 6.59 (d, J = 8.8 Hz, 2 H), 6.67 (d, J = 8.8 Hz, 2 H), 7.24–7.29 (m, 4 H), 7.34–7.39 (m, 3 H), 8.98 (s, 1 H) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 30.0, 55.3, 55.4, 112.9, 113.1, 114.2, 114.8, 126.6, 128.5, 128.6, 130.2, 130.8, 131.2, 131.3, 132.0, 138.8, 151.1, 152.3, 158.1, 159.0, 160.3 ppm. HRMS (ESI): calcd. for C₂₇H₂₄N₃O₂ [M + H]⁺ 422.1863; found 422.1864.

5-(4-Methoxyphenyl)-7-methyl-6-phenyl-4-[4-(trifluoromethyl)-phenyl]-7H-pyrrolo[2,3-d]pyrimidine (8i): The reaction was carried out by following general procedure E starting from the iodo derivative 7e (100 mg, 0.21 mmol) and 4-methoxyphenylboronic acid (34.9 mg, 0.23 mmol). The crude product was purified by silica gel column chromatography (CH₂Cl₂/acetone, 98:2 to 9:1) to afford 8i (71.4 mg, 0.16 mmol, 74%) as a fluorescent yellow solid (20% of starting material was also recovered). R_f = 0.5 (CH₂Cl₂/acetone, 8:2); m.p. 198–200 °C (acetone). IR (ATR diamond): $\tilde{\nu}$ = 3038, 2935, 2832, 1612, 1557, 1513, 1484, 1470, 1442, 1406, 1349, 1318, 1290, 1246, 1232, 1172, 1158, 1119, 1104, 1064, 1037, 1017, 953, 926, 895, 846, 804, 780, 763, 748, 734, 703 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 3.67 (s, 3 H), 3.84 (s, 3 H), 6.45 (d, J = 8.7 Hz, 2 H), 6.60 (d, J = 8.7 Hz, 2 H), 7.27–7.34 (m, 4 H), 7.36–7.40 (m, 5 H), 9.03 (s, 1 H) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 30.1, 55.2, 113.2, 113.8, 115.6, 124.2 (J = 273 Hz), 124.2 (J = 4 Hz), 125.8, 128.6, 128.9, 129.9, 130.3, 130.4 (J = 32 Hz), 131.0, 131.8, 139.5, 141.1, 151.2, 152.3, 157.7, 158.4 ppm. ¹⁹F NMR (376 MHz, CDCl₃): δ = –62.70 (CF₃) ppm. HRMS (ESI): calcd. for C₂₇H₂₁F₃N₃O [M + H]⁺ 460.1631; found 460.1632.

5,6-Bis(4-methoxyphenyl)-7-methyl-4-(3-tolyl)-7H-pyrrolo[2,3-d]pyrimidine (8j): The reaction was carried out by following general procedure E starting from the iodo derivative 7f (100 mg, 0.22 mmol) and 4-methoxyphenylboronic acid (36.7 mg, 0.24 mmol). The crude product was purified by silica gel column chromatography (CH₂Cl₂/acetone, 98:2 to 8:2) to afford 8j (54.6 mg, 0.13 mmol, 57%) as a beige solid (40% of starting material was also recovered). R_f = 0.20 (CH₂Cl₂/acetone, 9:1); m.p. 195–197 °C (acetone). IR (ATR diamond): $\tilde{\nu}$ = 3033, 2921, 2850, 1738, 1612, 1541, 1519, 1496, 1463, 1442, 1422, 1394, 1346, 1319, 1292, 1242, 1174, 1132, 1108, 1028, 961, 933, 917, 852, 835, 810, 798, 784, 741, 725, 709 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 2.04 (s, 3 H), 3.70 (s, 3 H), 3.81 (2s, 6 H), 6.48 (d, J = 8.9 Hz, 2 H), 6.66 (d, J = 8.9 Hz, 2 H), 6.89 (d, J = 8.9 Hz, 2 H), 7.00–7.06 (m, 3 H), 7.19 (d, J = 8.9 Hz, 2 H), 7.21–7.24 (m, 1 H), 8.99 (s, 1 H) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 21.1, 30.0, 55.3, 55.4, 113.0, 113.8, 114.0, 115.0, 122.7, 126.6, 126.7, 127.4, 129.3, 131.0, 131.9, 132.3, 137.0, 137.4, 138.9, 150.8, 152.2, 157.9, 159.1, 159.7 ppm. HRMS (ESI): calcd. for C₂₈H₂₆N₃O₂ [M + H]⁺ 436.2020; found 436.2020.

5-(4-Methoxyphenyl)-7-methyl-4-(3-tolyl)-6-[4-(trifluoromethyl)-phenyl]-7H-pyrrolo[2,3-d]pyrimidine (8k): The reaction was carried out by following general procedure E starting from the iodo derivative 7g (150 mg, 0.30 mmol) and 4-methoxyphenylboronic acid (50.8 mg, 0.36 mmol). The crude product was purified by silica gel column chromatography (CH₂Cl₂/acetone, 95:5) to afford 8k (84.5 mg, 0.18 mmol, 59%) as a white solid (34% of starting material was also recovered). R_f = 0.39 (CH₂Cl₂/acetone, 7:3); m.p. 191–193 °C (acetone). IR (ATR diamond): $\tilde{\nu}$ = 3055, 2955, 2923, 2854, 1708, 1557, 1539, 1514, 1438, 1416, 1349, 1319, 1287, 1243, 1224, 1179, 1120, 1086, 1033, 953, 894, 837, 809, 773, 763, 722, 701 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 2.05 (s, 3 H), 3.71 (s, 3 H), 3.83 (s, 3 H), 6.51 (d, J = 8.8 Hz, 2 H), 6.65 (d, J = 8.8 Hz, 2 H), 6.96–7.06 (m, 3 H), 7.21–7.25 (m, 1 H), 7.40 (d, J = 8.0 Hz, 2 H), 7.63 (d, J = 8.0 Hz, 2 H), 9.02 (s, 1 H) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 21.1, 30.2, 55.3, 113.3, 115.0, 115.3, 124.0 (J = 274 Hz), 125.5 (J = 4 Hz), 125.9, 126.7, 127.5, 129.6, 130.4, 130.7, 131.0, 131.5, 131.9, 134.6 (J = 1 Hz), 137.1, 137.2, 151.6, 152.5, 158.3, 160.2 ppm. ¹⁹F NMR (376 MHz, CDCl₃): δ = –62.73 (CF₃) ppm. HRMS (ESI): calcd. for C₂₈H₂₃F₃N₃O [M + H]⁺ 474.1788; found 474.1787. CCDC-963176 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

4-Chloro-5-(4-methoxyphenyl)-7-methyl-6-phenyl-7H-pyrrolo[2,3-d]pyrimidine (9): The reaction was carried out by following general procedure E starting from 10 (150 mg, 10a: 0.46 mmol or 10b: 0.41 mmol) and 4-methoxyphenylboronic acid (10a: 77.5 mg, 0.51 mmol or 10b: 68.4 mg, 0.45 mmol). The crude product was purified by silica gel column chromatography (CH₂Cl₂/acetone, 95:5) to afford 9 as a white solid in 33% (53.5 mg, 0.15 mmol from 10a) and 70% (99.2 mg, 0.28 mmol from 10b) yield. R_f = 0.35 (CH₂Cl₂/acetone, 95:5); m.p. 150–152 °C (acetone). IR (ATR diamond): $\tilde{\nu}$ = 2931, 2836, 1541, 1515, 1479, 1445, 1414, 1286, 1247, 1234, 1220, 1178, 1155, 1124, 1028, 955, 892, 834, 750, 708, 700 cm⁻¹. ¹H NMR (250 MHz, CDCl₃): δ = 3.81 (s, 3 H), 3.83 (s, 3 H), 6.83 (d, J = 8.8 Hz, 2 H), 7.19 (d, J = 8.8 Hz, 2 H), 7.25–7.34 (m, 2 H), 7.36–7.45 (m, 3 H), 8.70 (s, 1 H) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 33.4, 55.3, 113.1, 113.9, 115.8, 124.6, 128.6, 128.9, 129.8, 130.8, 132.9, 140.0, 150.3, 151.7, 151.8, 158.7 ppm. HRMS (ESI): calcd. for C₂₀H₁₇ClN₃O [M + H]⁺ 350.1055; found 350.1055.

5-Bromo-4-chloro-7-methyl-6-phenyl-7H-pyrrolo[2,3-d]pyrimidine (10a): A solution of 4-chloropyrimidine 5a (750 mg, 3.08 mmol) and NBS (639 mg, 3.59 mmol) in acetonitrile (5.5 mL) was transferred to a special microwave tube and irradiated in a microwave oven at 100 °C for 30 min. The progress of the reaction was monitored by TLC. After cooling, the solvent was removed under vacuo. Dichloromethane (20 mL) was added and the organic layer was washed with aqueous saturated Na₂S₂O₃ (2 × 16 mL), and NaOH (10%, 2 × 16 mL). Evaporation of the solvent under reduced pressure gave the crude product 10a (874.6 mg, 2.71 mmol, 87%) as a yellow solid, which did not require further purification. R_f = 0.41 (CH₂Cl₂/ethyl acetate, 9:1); m.p. 168–170 °C (ethyl acetate). IR (ATR diamond): $\tilde{\nu}$ = 3059, 2931, 1582, 1544, 1488, 1468, 1436, 1347, 1221, 1174, 1152, 1028, 962, 890, 780, 762, 701 cm⁻¹. ¹H NMR (250 MHz, CDCl₃): δ = 3.75 (s, 3 H), 7.47–7.58 (m, 5 H), 8.66 (s, 1 H) ppm. ¹³C NMR (63 MHz, CDCl₃): δ = 30.9, 87.3, 115.3, 128.7, 129.0, 130.0, 130.6, 140.8, 150.9, 151.3, 151.8 ppm. HRMS (ESI): calcd. for C₁₃H₁₀BrClN₃ [M + H]⁺ 321.9741; found 321.9742.

4-Chloro-5-iodo-7-methyl-6-phenyl-7H-pyrrolo[2,3-d]pyrimidine (10b): The reaction was carried out by following general procedure D starting from 4-chloropyrimidine (5a) (300 mg, 1.23 mmol, 1.0 equiv.) to afford 10b (435.5 mg, 1.18 mmol, 96%) as a yellow solid. *R*_f = 0.34 (CH₂Cl₂/ethyl acetate, 95:5); m.p. 203–205 °C (ethyl acetate). IR (ATR diamond): $\tilde{\nu}$ = 3744, 3058, 2924, 2854, 1541, 1479, 1434, 1343, 1265, 1217, 1168, 1026, 952, 886, 765, 702 cm^{–1}. ¹H NMR (400 MHz, CDCl₃): δ = 3.73 (s, 3 H), 7.41–7.47 (m, 2 H), 7.53–7.58 (m, 3 H), 8.64 (s, 1 H) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 31.4, 54.0, 117.5, 129.0, 130.1, 130.5, 130.8, 144.8, 150.6, 152.2, 152.4 ppm. HRMS (ESI): calcd. for C₁₃H₁₀ClIN₃ [M + H]⁺ 369.9603; found 369.9604.

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672 [1] U. Battaglia, J. E. Long, M. S. Searle, C. J. Moody, *Org. Biomol. Chem.* 2011, 9, 2227–2232.

673 [2] S. Ding, T. Y. H. Wu, A. Brinker, E. C. Peters, W. Hur, N. S. Gray, P. G. Schultz, *Proc. Natl.*

674 *Acad. Sci. USA* 2003, 100, 7632–7637.

675 [3] X. Y. Jiao, D. J. Kopecky, J. S. Liu, J. Q. Liu, J. C. Jaen, M. G. Cardozo, R. Sharma, N. Walker,

676 H. Wesche, S. Li, E. Farrelly, S. H. Xiao, Z. Wang, F. Kayser, *Bioorg. Med. Chem. Lett.* 2012,

677 22, 6212–6217.

678 [4] a) J. E. Chin, C. A. Hatfield, G. E. Winterrowd, R. F. Krzesicki, K. L. Shull, I. M. Richards,

679 *JPEN J. Parenter. Enteral Nutr.* 1999, 290, 188–195; b) M. F. Jarvis, H. Yu, B. F. Cox, J.

680 Polakowski, *Pain* 2002, 96, 107–118.

681 [5] N. Chakka, H. Bregman, B. Du, H. N. Nguyen, J. L. Buchanan, E. Feric, J. Ligutti, D. Liu, J. S.

682 McDermott, A. Zou, S. I. McDonough, E. F. DiMauro, *Bioorg. Med. Chem. Lett.* 2012, 22,

683 2052–2062.

684 [6] M. Vrabel, P. Horakova, H. Pivokova, L. Kalachova, H. Cernocka, H. Cahova, R. Pohl, P.

685 Sebest, L. Havran, M. Hocek, M. Fojta, *Chem. Eur. J.* 2009, 15, 1144–1154.

686 [7] F. Seela, H. Steker, *Liebigs Ann. Chem.* 1984, 1719–1730.

687 [8] a) S. Tumkevicius, J. Dodonova, K. Kazlauskas, V. Masevicius, L. Skardziute, S. Jursenas,

688 *Tetrahedron Lett.* 2010, 51, 3902–3906; b) S. Tumkevicius, J. Dodonova, *Synlett* 2011, 1705–

689 1708; c) M. Vrábel, R. Pohl, I. Votruba, M. Sajadi, S. A. Kovalenko, N. P. Ernsting, M. Hocek,

690 *Org. Biomol. Chem.* 2008, 6, 2852–2860.

691 [9] a) V. P. Kumar, K. M. Frey, Y. Wang, H. K. Jain, A. Gangjee, K. S. Anderson, *Bioorg. Med.*

692 *Chem. Lett.* 2013, 23, 5426–5428; b) A. Gangjee, N. Zaware, S. Raghavan, J. Yang, J. E.

693 Thorpe, M. A. Ihnat, *Bioorg. Med. Chem.* 2012, 20, 2444–2454; c) L. Wang, C. Cherian, S. K.

694 Desmoulin, S. Mitchell-Ryan, Z. Hou, L. H. Matherly, A. Gangjee, *J. Med. Chem.* 2012, 55,

695 1758–1770.

696 [10] a) E. C. Taylor, B. Liu, *J. Org. Chem.* 2003, 68, 9938–9947; b) CAS registration entry with

697 query 4-X, 6-C, 5,7-H-pyrrolo[2,3-d]pyrimidine: 1345 for X = N (January 2010).

- 698 [11] a) Y. Liu, J. Fang, H. Cai, F. Xiao, K. Ding, Y. Hu, *Bioorg. Med. Chem.* 2012, 20, 5473–5482;
699 b) J. L. Henderson, S. M. McDermott, S. L. Buchwald, *Org. Lett.* 2010, 12, 4438–4441.
- 700 [12] M. H. Jung, H. Kim, W. K. Choi, M. I. El-Gamal, J. H. Park, K. H. Yoo, T. B. Sim, D. Baek, J.
701 M. Hah, J. H. Cho, C. H. Oh, *Bioorg. Med. Chem. Lett.* 2009, 19, 6538–6543.
- 702 [13] M. S. Mohamed, A. E. Rashad, M. Abdel-Momem, S. S. Fatahalla, *Z. Naturforschung, Teil C*
703 2007, 62, 27–31.
- 704 [14] K. M. H. Hilmy, M. M. A. Khalifa, A. A. Hawata, R. M. A. Keshk, A. A. El-Torgman, *Eur. J.*
705 *Med. Chem.* 2010, 45, 5243–5250.
- 706 [15] H. S. Choi, Z. Wang, W. Richmond, X. He, K. Yang, T. Jiang, T. Sim, D. Karanewsky, X. Gu,
707 V. Zhou, Y. Liu, O. Ohmori, J. Caldwell, N. Gray, Y. He, *Bioorg. Med. Chem. Lett.* 2006, 16,
708 2173–2176.
- 709 [16] S. H. Spergel, D. R. Okoro, W. Pitts, *J. Org. Chem.* 2010, 75, 5316–5319.
- 710 [17] A. Mayasundari, N. Fujii, *Tetrahedron Lett.* 2010, 51, 3597–3598.
- 711 [18] A. Carpita, A. Ribecai, P. Stabile, *Tetrahedron* 2010, 66, 7169–7178.
- 712 [19] J. Dodonova, L. Skardziute, K. Kazlauskas, S. Jursenas, S. Tumkevicius, *Tetrahedron* 2012, 68,
713 329–339.
- 714 [20] L. C. W. Chang, R. F. Spanjersberg, J. K. von F. D. Künzel, T. Mulder-Krieger, J. Brussee, A.
715 P. Ijzerman, *J. Med. Chem.* 2006, 49, 2861–2867.
- 716 [21] a) S. Takahashi, Y. Kuroyama, K. Sonogashira, N. Hagihara, *Synthesis* 1980, 627–630; b) K.
717 Sonogashira, in: *Comprehensive Organic Synthesis* (Eds.: B. M. Trost, I. Fleming), Pergamon,
718 New York 1991, vol. 3, p. 521–561; c) R. Chinchilla, C. Nájera, *Chem. Soc. Rev.* 2011, 40,
719 5084–5121; d) M. Schilz, H. Plenio, *J. Org. Chem.* 2012, 77, 2798–2807.
- 720 [22] For general examples of microwave-assisted reactions, see: a) E. Buxaderas, D. A. Alonso, C.
721 Nájera, *Eur. J. Org. Chem.* 2013, 5864–5870; b) G. Broggini, V. Barbera, E. M. Beccalli, U.
722 Chiacchio, A. Fasana, S. Galli, S. Gazzola, *Adv. Synth. Catal.* 2013, 355, 1640–1648; c) H. H.
723 Nguyen, M. J. Kurth, *Org. Lett.* 2013, 15, 362–365; d) W. Qian, L. Zhang, H. Sun, H. Jiang, H.
724 Liu, *Adv. Synth. Catal.* 2012, 354, 3231–3236; e) M. Baghbanzadeh, C. Pilger, C. O. Kappe, J.
725 *Org. Chem.* 2011, 76, 8138–8142; f) J. F. Cívicos, D. A. Alonso, C. Nájera, *Adv. Synth. Catal.*

726 2011, 353, 1683–1687; g) E. M. Beccalli, A. Bernasconi, E. Borsini, G. Broggini, M.
 727 Rigamonti, G. Zecchi, *J. Org. Chem.* 2010, 75, 6923–6932; h) D. Liptrot, L. Alcaraz, B.
 728 Roberts, *Adv. Synth. Catal.* 2010, 352, 2183–2188.
 729 [23] a) M. Hocek, A. Holý, I. Votruba, H. Dvůřáková, *J. Med. Chem.* 2000, 43, 1817–1825; b) N. T.
 730 S. Phan, M. van Der Sluys, C. W. Jones, *Adv. Synth. Catal.* 2006, 348, 609–679; c) S. Asano, S.
 731 Kamioka, Y. Isobe, *Tetrahedron* 2012, 68, 272–279; d) A. Omumi, D. G. Beach, M. Baker, W.
 732 Gabryelski, R. A. Manderville, *J. Am. Chem. Soc.* 2011, 133, 42–50.
 733 [24] a) Ü. Yilmaz, S. Deniz, H. Hüçükbay, N. Sireci, *Molecules* 2013, 18, 3712–3724; b) A. El
 734 Akkaoui, S. Berteina-Raboin, A. Mouaddib, G. Guillaumet, *Eur. J. Org. Chem.* 2010, 862–871;
 735 c) J. Koubachi, A. El Kazzouli, S. Berteina-Raboin, A. Mouaddib, G. Guillaumet, *J. Org. Chem.*
 736 2007, 72, 7650–7655.
 737 [25] M. Singer, A. Jäschke, *J. Am. Chem. Soc.* 2010, 132, 8372–8377.
 738 [26] C. W. van der Westhuyzen, A. L. Rousseau, C. J. Parkinson, *Tetrahedron* 2007, 63, 5394–5405.
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Legends to figures

Figure 1. Purines and 7-deazapurines.

Scheme 1. Two synthetic routes to the triarylalted pyrrolopyrimidines 8.

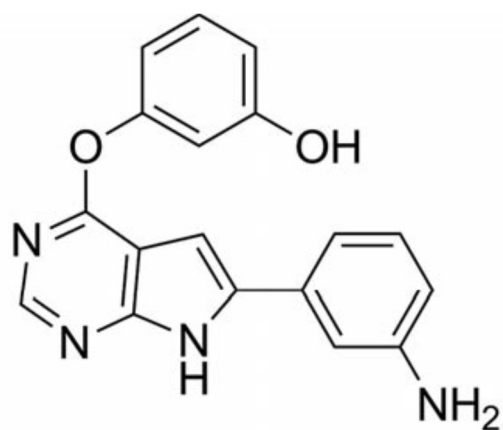
Scheme 2. Alkynylation of iodopyrimidines 3.

Scheme 3. Intramolecular cyclization of diarylalkynyl derivatives 4.

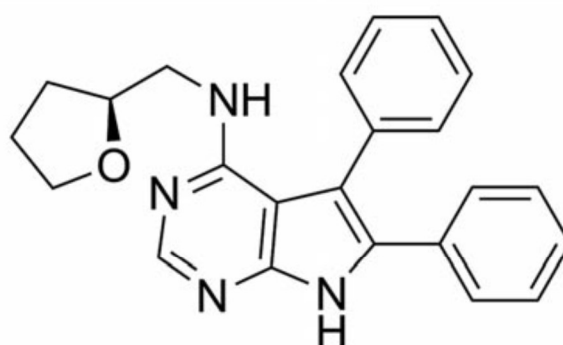
Scheme 4. Diarylation of pyrrolopyrimidines.

Figure 2. Molecular structure of 8k. Displacement ellipsoids are drawn at the 30% probability level and H atoms are represented by circles of arbitrary radii.

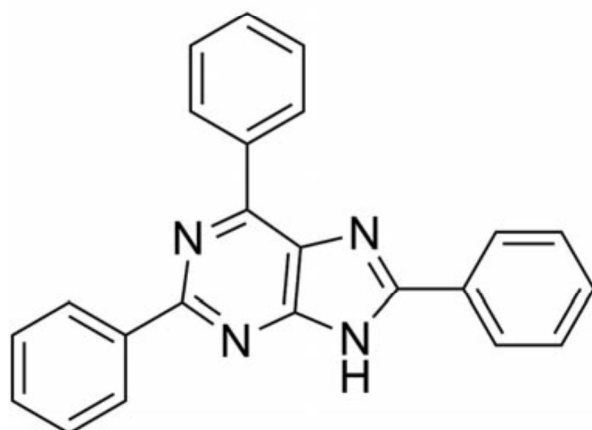
FIGURE 1



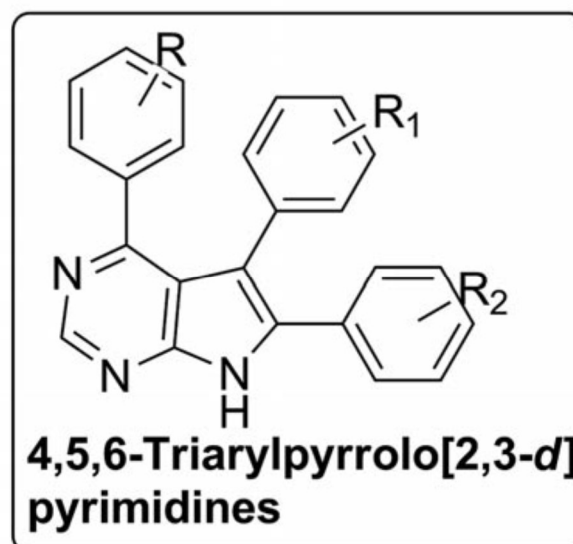
TWS119, GSK-3 β inhibitor



Pyrrolo[2,3-*d*]pyrimidines
ACK1 inhibitors

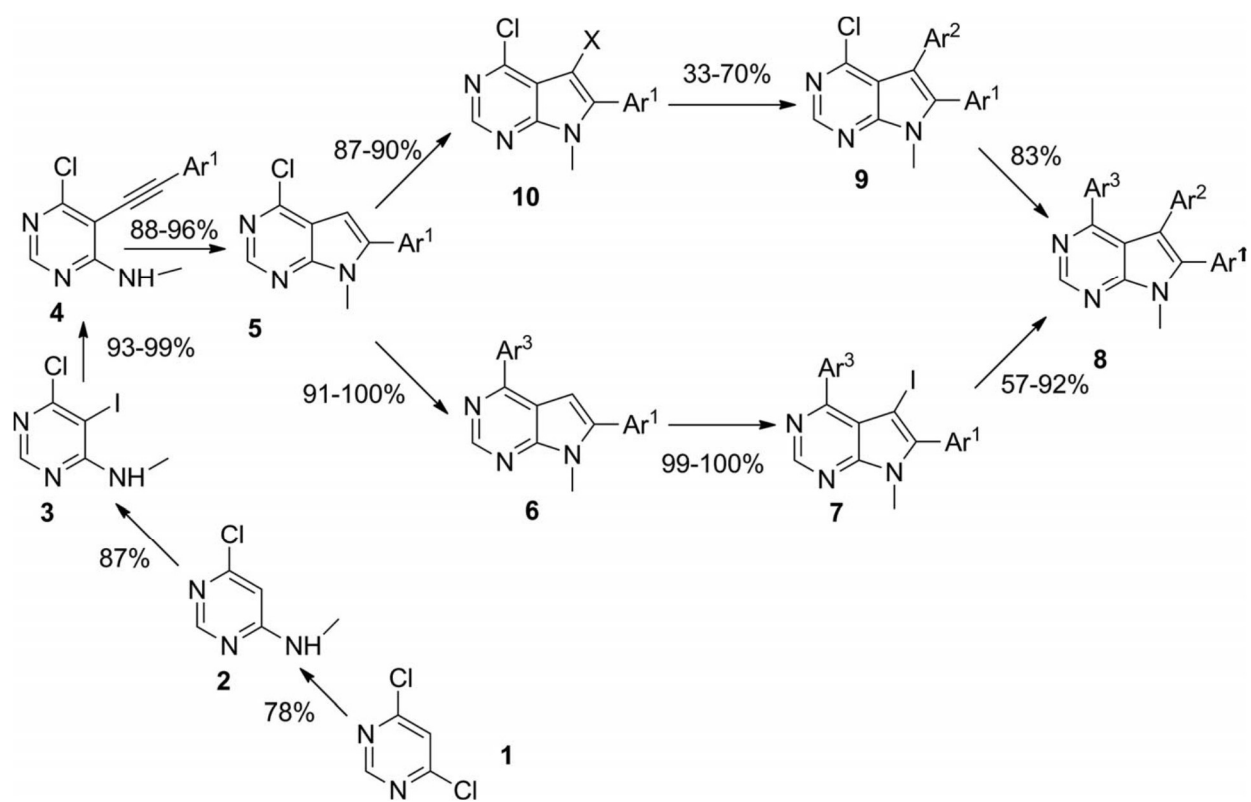


2,6,8-Triarylpyrrolo[2,3-*d*]pyrimidines
Adenosine receptor antagonists

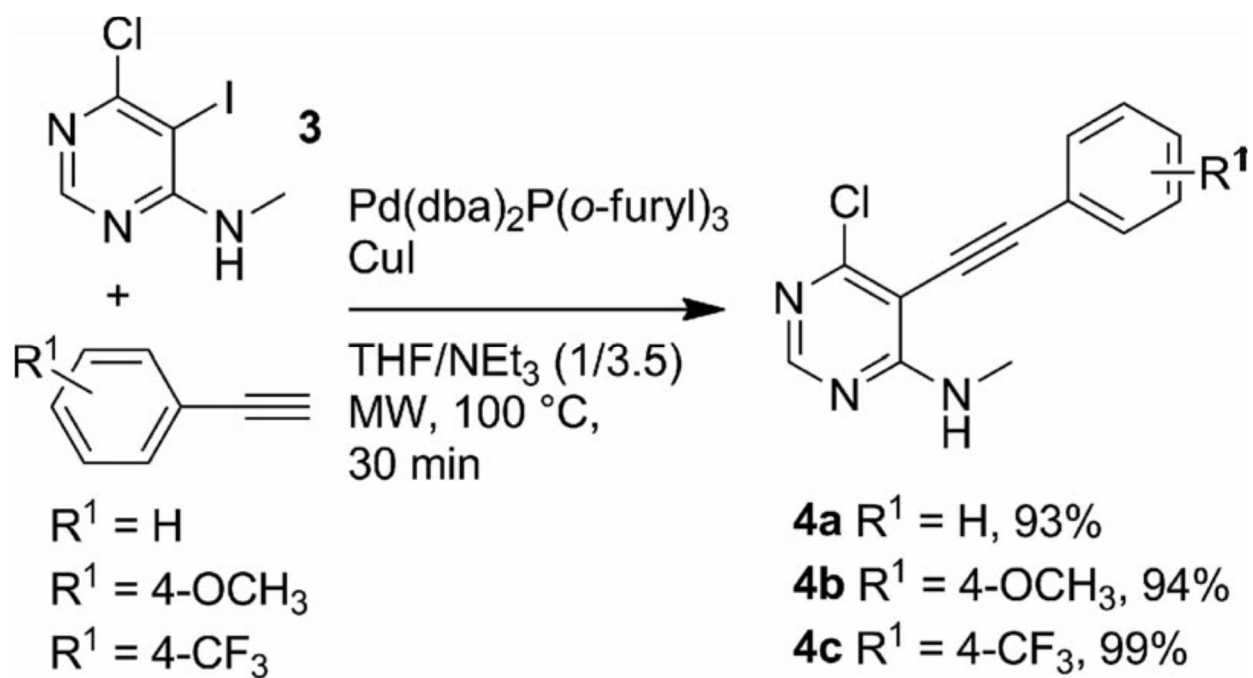


4,5,6-Triarylpyrrolo[2,3-*d*]pyrimidines

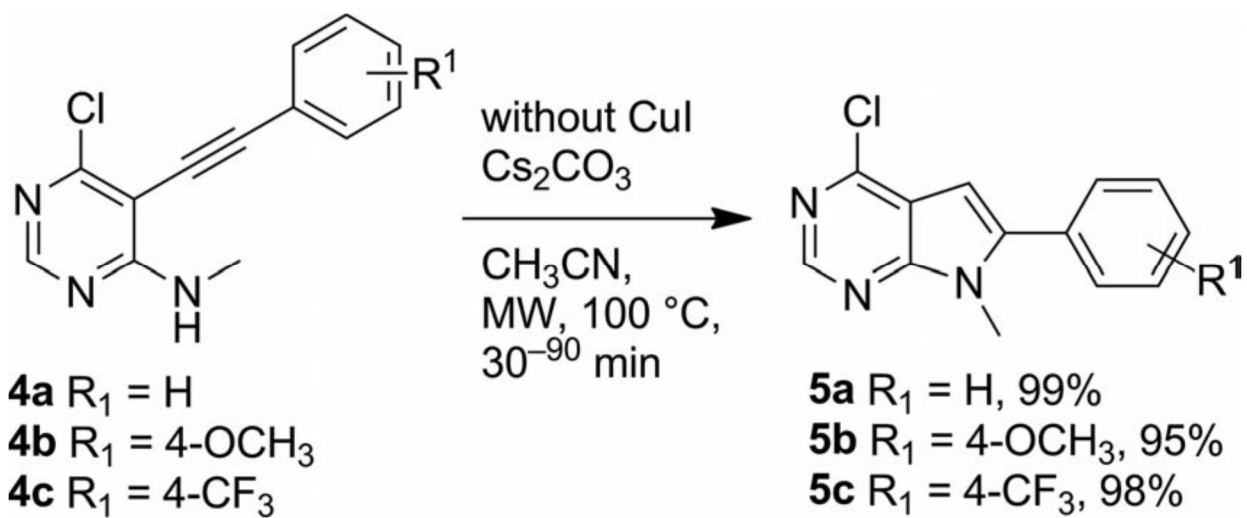
SCHEME 1



SCHEME 2



SCHEME 3



SCHEME 4

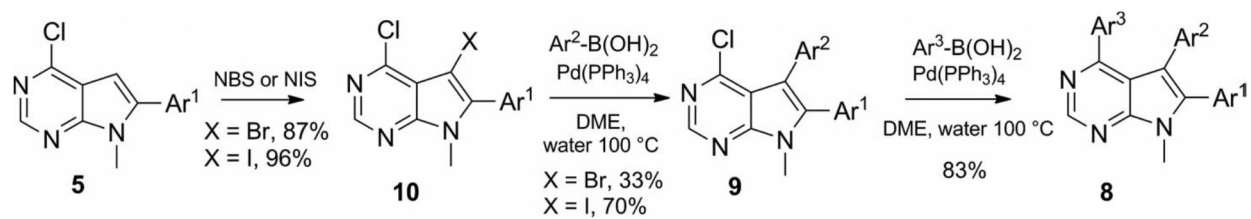


FIGURE 2

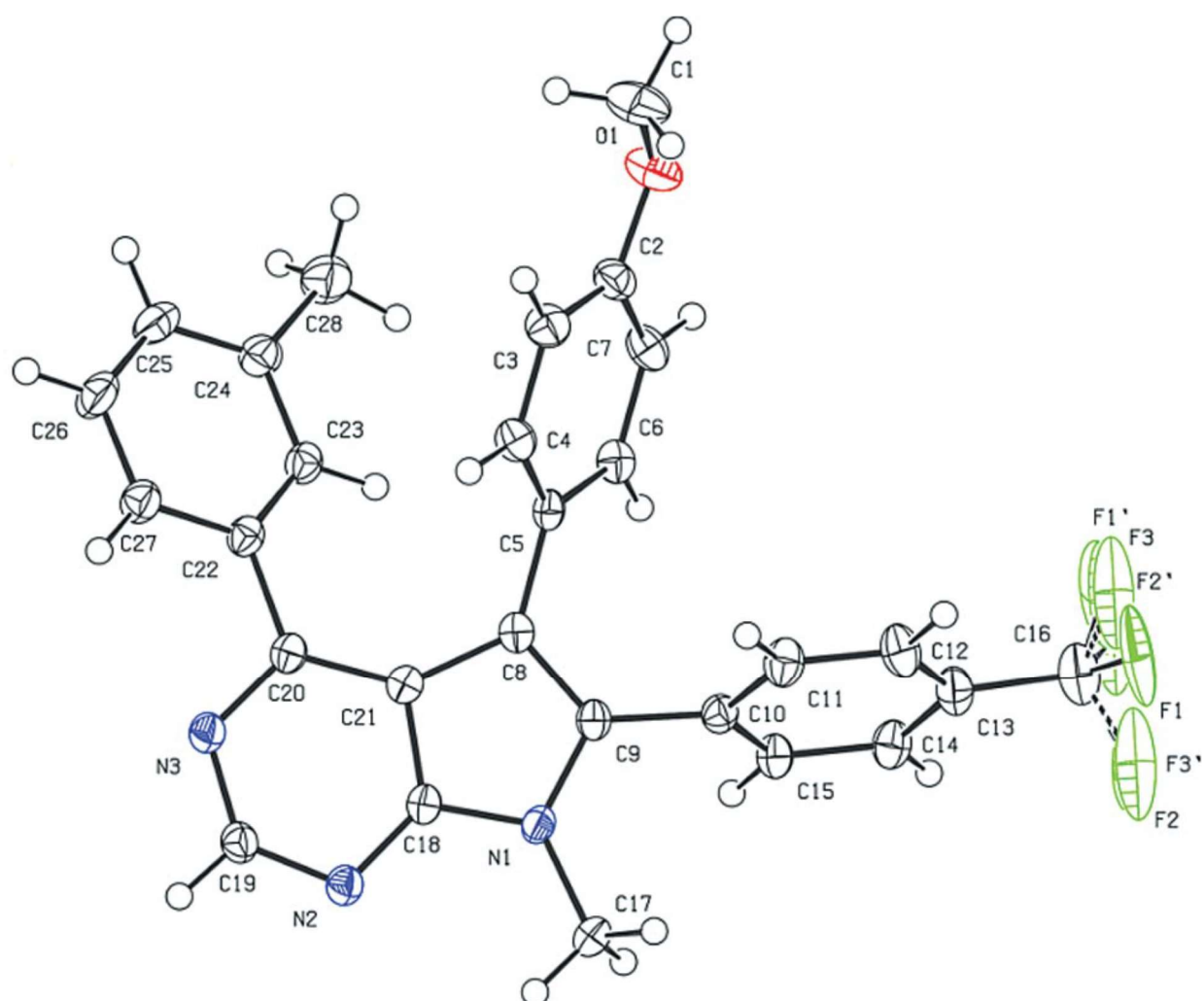


Table 1. Suzuki–Miyaura cross-coupling reactions at the 4-position of heterocycle 5 leading to 4,6-disubstituted pyrrolopyrimidines 6.

Entry	R ¹	R ³	Product	Yield (%) ^[a]
1	H	3-Me		91
2	H	2-Me		99
3	H	4-Me		96
4	H	4-MeO		100
5	H	4-CF ₃		98
6	4-MeO	3-Me		96
7	4-CF ₃	3-Me		96

[a] Yield of isolated compound.

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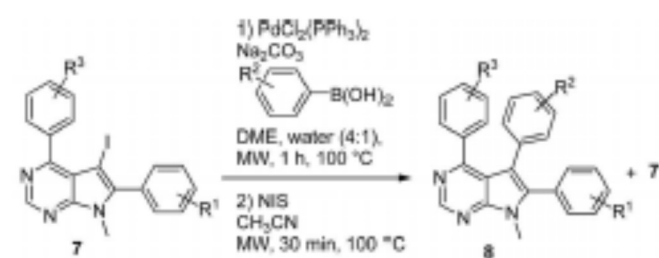
Table 2. Iodination of pyrrolopyrimidines 6.



Entry	R ¹	R ³	Product	Yield (%) ^[a]
1	H	3-Me		100
2	H	2-Me		99
3	H	4-Me		100
4	H	4-MeO		100
5	H	CF ₃		100
6	4-MeO	3-Me		100
7	4-CF ₃	3-Me		100

[a] Yield of isolated compound.

805 **Table 3.** Suzuki–Miyaura cross-coupling reaction at C-5 of pyrrolopyrimidines 7..
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Entry	Compd.	R ¹	R ²	R ³	Product	Yield ^[a]
1	7a	H	4-MeO	3-Me		88
2	-	H	4-CF ₃	3-Me		67
3	-	H	4-Me	3-Me		71
4	-	H	3-Me	3-Me		76
5	-	H	2-Me	3-Me		traces
6	7b	H	4-MeO	2-Me		94
7	7c	H	4-MeO	4-Me		65
8	7d	H	4-MeO	4-MeO		92
9	7e	H	4-MeO	4-CF ₃		74
10	7f	4-MeO	4-MeO	3-Me		57
11	7g	4-CF ₃	4-MeO	3-Me		59

[a] Yield of isolated compound.